



ORIGINAL RESEARCH ARTICLE

DESIGN, SYNTHESIS AND SPECTRAL CHARACTERIZATION OF DARUNAVIR ANALOGUEHanumantharao Bommena^{1,2}, Subramanyeswara Rao Venkata Inti¹, Ravikanth Vysyaraju¹, Bala Murali Krishna² and Syama Sundar Bethanabatlal*²¹Chemical Research and Development, Aurobindo pharma Ltd., Indrakaran (V), Sangareddy (M), Medak Dist- 502329. Andhra Pradesh, India.²Department of chemistry, Acharya Nagarjuna University, Nagarjuna nagar, Guntur-522510, Andhra Pradesh, India.

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Abstract: The stereo selective novel synthesis of (3S,3aR,6aS)-Hexahydrofuro [2,3-b] furan-3-yl (1S,2R)-3-(4-amino-N-isobutylphenylsulfonamido)-1-(4-fluorophenyl)-2-hydroxypropylcarbamate (1), it involves the eight synthetic steps mainly which was accomplished from the intermediate 4-Amino-N-((2R,3S)-3-amino-3-(4-fluorophenyl)-2-hydroxypropyl)-N-isobutyl benzenes sulfonamide. This reaction gave 88% yield, with an optical purity of 99.5% and purity of 99% for chiral alcohol.

Key words: Stereoselective, L-phenyl glycine, Carbamate, Biological properties

INTRODUCTION

AIDS (Acquired immunodeficiency syndrome) is a chronic, life-frightening ailment caused by the human immunodeficiency virus (HIV). HIV, which in turn interferes with the body's ability to fight effectively against viruses, bacteria and fungi that cause the disease by destroying the cells of the immune system. HAART (highly active antiretroviral therapy) introduced in 1996, combined with HIV-1 protease inhibitors and reverse transcriptase inhibitors has theatrically transformed the management of HIV/AIDS.¹ For current drug regimens to treat HIV infection, HIV protease inhibitors are essential components. By interfering with an enzyme known as HIV protease, HIV replication is interrupted by protease inhibitors at a later stage in its life cycle by interfering with an enzyme known as HIV protease.² The innovative therapeutic treatments and regimens frequently change the impact of disease. HIV-1 protease inhibitors are reported as the most potent anti-AIDS drugs to date and are necessary constituents of HAART.³⁻⁵ Anti-AIDS chemotherapy based on HIV-1 protease and reverse-transcriptase inhibitors has been extremely effective in decreasing the mortality rates in HIV-infected patients. HIV protease inhibitors, for example indinavir, darunavir, amprenavir, atazanavir, nelfinavir, and saquinavir⁶⁻¹¹ all belong to the HEA class of inhibitors,¹² and synthesis of this class of inhibitors are reorted.¹³ HIV-1 protease displays a critical role in the virus life cycle by processing the viral Gag and Gag-Pol poly proteins into structural and functional proteins necessary for viral maturation.

Antiretroviral therapy with HIV protease inhibitor drugs (PI) has been associated with abnormalities in carbohydrate metabolism including insulin resistance, hyperglycemia and development of diabetes mellitus in patients with HIV infection.^{14, 15}

Although the pathogenesis of glucose abnormalities during HIV infection is believed to be multifactorial, two independent short term studies of healthy HIV-seronegative volunteers showed that induction of insulin resistance is a direct effect of treatment with the PI indinavir (IDV).^{16,17} The fixed-combination PI lopinavir/ritonavir (LPV/r) has also been shown to induce insulin resistance, as assessed by oral glucose tolerance testing in healthy volunteers.¹⁸ Although induction of insulin resistance by IDV preceded any significant changes in plasma lipids, impairment in glucose tolerance with LPV/r was temporarily associated with increase in free fatty acids and triglycerides. It is not clear whether metabolic differences in insulin action by a PI primarily reflect direct effects on glucose uptake or represent consequences of an effect on adipose tissue. However, the earliest and most direct PI effect on carbohydrate metabolism appears to be on insulin-mediated glucose uptake, as insulin resistance is apparent even after a single dose of IDV.¹⁹ Data from *in vitro* models of glucose uptake suggest that PI contribute to insulin resistance by inhibiting the insulin-regulated glucose transporter GLUT4.²⁰ Several investigators have independently confirmed inhibition of glucose uptake *in vitro* in various models of glucose uptake, including human adipocytes.^{21,22} Previous studies from our laboratory and others have suggested that phosphorylated chemical entities having diversified biological properties including anti cancer^{23, 24}, antiviral²⁵, antimicrobial^{26, 27} and anti-Alzheimer's.²⁸

We herein report our efforts on this synthetic strategy and ultimately achieved a relatively short synthesis with high stereo selective darunavir analogue.

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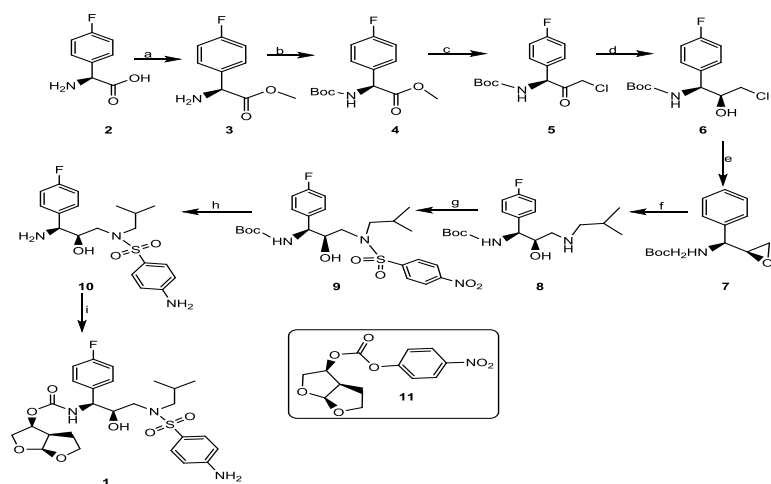
MATERIALS AND METHODS

All the raw materials were procured from Merck laboratories and reagents and solvents were used without further purification.

RESULTS AND DISCUSSION

Herein, for the first time we report a new and an efficient method for the highly stereo selective synthesis of darunavir analogue from the readily available L-phenyl glycine. Our synthetic strategy involves esterification of L-phenyl glycine **2**, stereo selective reduction of α -haloketone **5** followed by S_N2 ring opening of chiral (*S,S*)-epoxide **7** with isobutyl amine. Finally the darunaviranalogue **1** was achieved from enantiomeric pure amino alcohol **10** and bis THF phenyl carbonate **11** (Scheme 1).

Synthesis of (3*S*, 3*aR*, 6*aS*)-Hexahydrofuro [2,3-*b*]furan-3-yl (1*S*,2*R*)-3-(4-amino-N-isobutylphenylsulfonamido)-1-(4-fluorophenyl)-2-hydroxypropylcarbamate **1** was achieved by the esterification of L-phenyl glycine **2** using thionyl chloride under reflux conditions. The amino group of methyl ester of L-phenyl glycine **2** was protected by (Boc)₂O. The ester **3** was treated with diisopropylamido magnesium chloride in THF and α -chloroacetic acid to obtain α -chloro ketone **4**. The diastereo selective reduction of α -chloro ketone **4** was achieved by using aluminum isopropoxide in isopropyl alcohol to get α -chlorohydrin **5** with 99% ee. The (*S,S*)-epoxide **7** was acquired by base-induced epoxide formation of α -chlorohydrin **6**. Then S_N2 ring opening of chiral (*S,S*)-epoxide **7** with isobutyl amine under solvent free conditions followed by *in situ* reaction of *p*-nitro benzene sulfonyl chloride yielded the intermediate compound **9**. The nitro group present in amido alcohol **9** is reduced to amine by Fe-HCl followed by deprotection of the Boc group afforded the required amino alcohol **10**. Finally, the stereo selective synthesis of darunaviranalogue **1** was achieved by reacting amino alcohol **10** with (3*R*,3*aS*,6*aR*)-hexahydrofuro [2,3-*b*]furan-3-yl 4-nitrophenyl carbonate **11**. Analogue of darunavir **1** with enantiomeric ratio of 99% was accomplished. The chemical structures of all the compounds were characterized by IR, ¹H, ¹³C, and HRMS studies and their data are presented in the experimental section. Characteristic IR stretching absorptions were observed in the regions 3377-3345 and 3555-3452 cm⁻¹ for N-H, O-H, respectively.²⁹⁻³³ In the ¹H NMR spectra of the chemical shifts of aromatic hydrogens of the phenyl ring appeared as multiplets in the region δ 7.11-7.36.³⁴⁻³⁷ In ¹³C NMR chemical shifts for compounds were observed in their expected regions.



Scheme 1

Reaction Conditions: (a). SOCl₂, MeOH, reflux 5 h, (b). (Boc)₂O, sat. NaHCO₃ solution, rt, 16 h, 88% (c). MeMgCl, ClCH₂CO₂H, Diisopropyl amine, THF, HCl, 80%; (d). Al(OCHMe₂)₃, IPA, reflux, 3 h, 90%; (e). 20% NaOH Solution, Methanol, 20 °C, 90 min, 85%; (f). Isobutyl amine, ethyl acetate; (g). *p*-nitro benzene sulfonyl chloride, TEA, Chloroform, 85%; (h). Fe/HCl, Conc.HCl, 96%; (i). Methylene dichloride, (3*R*,3*aS*,6*aR*)-hexahydrofuro [2,3-*b*]furan-3-yl 4-nitrophenyl carbonate (**10**), rt, 8 h, 93%.

EXPERIMENTAL SECTION

Material section

Chemicals were procured from Sigma-Aldrich, Merck and Lancaster, and were used as such without further purification. Melting points (m.p.) were determined using a calibrated thermometer by Guna Digital Melting Point apparatus. They expressed in degrees centigrade (°C) and are uncorrected. Infrared Spectra (IR) were obtained on a Perkin-Elmer Model 281-B spectrophotometer. Samples were analyzed as potassium bromide (KBr) disks. Absorptions were reported in wave numbers (cm⁻¹). ¹H and ¹³C NMR spectra were recorded as solutions in DMSO-*d*₆ on a Bruker AMX 400 MHz spectrometer operating at 400 MHz for ¹H and 100 MHz for ¹³C. The ¹H and ¹³C chemical shifts were expressed in parts per million (ppm) with reference to tetramethylsilane (TMS). Optical rotations (in degrees, °) were recorded in methanol on a Perkin-Elmer Model 241 polarimeter at the sodium D line. APCI mass spectra were recorded on a Jeol SX 102 DA / 600 Mass spectrometer. Elemental analyses were performed by Central Drug Research Institute, Lucknow, INDIA.

Synthesis

(*S*)-Methyl 2-(tert-butoxycarbonylamino)-2-(4-fluorophenyl) acetate (**4**) was prepared from L-phenyl glycine **2** according to reported method.³⁸

Synthesis of (S)-tert-butyl 3-chloro-1-(4-fluorophenyl)-2-oxopropylcarbamate (5):³⁸⁻⁴¹

Part-A: Diisopropylamine (22.4 g, 0.2 mol) was added to a refluxed solution of methyl magnesium chloride (74.5 mL) in THF (3N solution) under nitrogen atmosphere. The reaction mixture was stirred for 1h at ambient temperature and it was cooled to room temperature.

Part-B: Meanwhile, monochloro acetic acid (7.08 g, 0.0749 mol) in tetrahydrofuran (7 mL) was taken into a 250 mL 4-neck round bottomed flask. Methyl magnesium chloride (28 mL) in THF (3N solution) was added at 10 °C under N₂ atmosphere. (S)-Methyl-2-(tert-butoxycarbonylamino)-2-(4-fluorophenyl) acetate (14 g, 0.049mol) in THF was added to the reaction mixture during 30 min at 10°C, later diisopropylamido magnesium chloride (Part-A) complex was added below 10°C during 45 min, then the reaction mixture was refluxed for 2 h. Progress of the reaction was monitored by thin layer chromatography (hexane: ethyl acetate, 7:3), after completion of the reaction, it was cooled to room temperature, quenched into water, pH was adjusted to 2.5 with conc. HCl and separated the two layers, aqueous layer was extracted with ethyl acetate (2x50 mL). The total extract was washed with saturated aqueous NaHCO₃ solution and 10% aqueous NaCl solution, organic layer was dried over Na₂SO₄ and concentrated in a rotary evaporator to obtain the crude product, it was recrystallised from n-hexane to get the light pink colored solid (3) (12.22 g, 82 %); mp 130-133 °C. R_f0.42; (hexane: ethyl acetate, 7:3 v/v): IR (KBr). 3345, 2965, 2341, 1748, 1686, 1566, 1300, 1264, 1166, 711 cm⁻¹. ¹H NMR (400 MHz, DMSO-d₆) δ 7.28-7.36 (m, 2H), 7.14 (d, J = 6.8 Hz, 2H), 5.08 (d, J = 5.6 Hz, 1H), 4.61 (q, 1H), 4.17 (d, J = 16.4 Hz, 1H), 3.88 (d, J = 16.0 Hz, 1H), 1.36 (s, 9H) ppm. ¹³C NMR: 176.2, 151.5, 131.4, 127.2, 125.1, 121.3, 81.2, 69.5, 41.5, 36.5, 25.6. $[\alpha]_D^{25} = +35.8^\circ$ (c = 1 in MeOH). LCMS (m/z): 303 [M+H]⁺.

Synthesis of tert-butyl (1S,2S)-3-chloro-1-(4-fluorophenyl)-2-hydroxypropylcarbamate (6):^{38,42} To a stirred solution of (S)-Tert-butyl 3-chloro-1-(4-fluorophenyl)-2-oxopropylcarbamate (5) (9.5 g, 0.03 mol) in 20 mL of isopropanol, aluminum isopropoxide (3.32 g, 0.02 mol) in isopropyl alcohol (95 mL) was added at room temperature. The resulting mixture was refluxed for 3 h, after completion of the reaction half of the isopropyl alcohol was concentrated in a rotary evaporator, and then water was added, pH was adjusted to 3.5 with glacial acetic acid at 30 °C. Further the reaction mixture was stirred for 2 h at room temperature, pink colored solid was separated and it was filtered, washed with water, dried under vacuum at 45 °C to get the light pink colored solid (4) (8.17

g, 86%); mp 155-157 °C. R_f0.52; (hexane: ethyl acetate, 7:3 v/v): IR (KBr). 3356, 2970, 1682, 1535, 1438, 1156, 1001, 695 cm⁻¹. ¹H NMR (400 MHz, DMSO-d₆) δ 7.18-7.28 (m, 2H), 7.11 (d, J = 6.8 Hz, 2H), 6.56 (d, J = 8.4 Hz, 1H), 5.44 (d, J = 6.0 Hz, 1H), 3.55-3.66 (m, 3H), 3.59 (q, 1H), 2.89 (d, J = 2.4 Hz, 1H), 2.79 (d, J = 9.6 Hz, 1H), 1.37 (s, 9H) ppm. ¹³C NMR: 160.5, 136.8, 129.1, 126.3, 124.5, 82.1, 79.2, 53.5, 46.8, 38.5, 26.6. $[\alpha]_D^{25} = +16.5^\circ$ (c = 1 in MeOH). LCMS (m/z): 305 [M+H]⁺.

Synthesis of tert-butyl formate (S, S)-oxiran-2-yl(phenyl)methanamine (7): To a stirred solution of tert-butyl (1S,2S)-3-chloro-1-(4-fluorophenyl)-2-hydroxypropylcarbamate (6) (7 g, 0.02 mol) in Methanol (56 mL), 20% NaOH solution (14 mL) was added at 20°C and continued stirring for 90 min. Progress of the reaction was monitored by TLC (hexane: ethyl acetate 7:3), then the organic layer was taken into a reaction flask and heated to 45 °C, water was added slowly at ambient temperature, the resulting mixture was cooled to room temperature, stirred for 2 h, the solid that separated was filtered and washed with water, product was dried under vacuum at 45°C to get white colored solid (5) (5.95 g, 85 %); mp 123-126 °C. R_f0.45; (hexanes: ethyl acetate, 7:3 v/v): IR (KBr). 3377, 2981, 1680, 1524, 1170, 928, 605 cm⁻¹. ¹H NMR (400 MHz, DMSO-d₆) δ 7.21-7.33 (m, 5H), 4.43 (s, 1H), 3.69 (s, 1H), 2.76-2.99 (m, 5H), 1.38 (s, 9H) ppm. ¹³C NMR: 156.3, 138.2, 128.5, 127.6, 125.2, 80.3, 61.4, 56.6, 40.1, 38.4, 26.5. $[\alpha]_D^{25} = +17.6^\circ$ (c = 1 in acetonitrile) LCMS (m/z): 250 [M+H]⁺.

Synthesis of tert-butyl (1S,2R)-1-(4-fluorophenyl)-2-hydroxy-3-(N-isobutyl-4-nitrophenylsulfonamido) propylcarbamate (9):^{38,43-50} tert-butyl formate (S, S)-oxiran-2-yl(phenyl) methanamine (7) (4.6 g, 0.02 mol) in isobutyl amine (13.8 g, 0.12 mol) was refluxed for 3 h, reaction progress was monitored by TLC (hexane: ethyl acetate, 7:3), excess isobutyl amine was completely removed under reduced pressure. A white colored solid was obtained and it was cooled to room temperature, chloroform (60 mL) and triethylamine (2.3 g, 0.02 mol) were added, the resulting mixture was raised to reflux, p-nitro benzenesulphonyl chloride (4.0 g, 0.02 mol) in chloroform (25 mL) was added to the above reaction mixture at ambient temperature during 2 h, the resulting reaction mixture was stirred for 2 h. Reaction progress was monitored by TLC (hexane: ethyl acetate, 7:3), then the reaction mixture was cooled to room temperature, further it was washed with water, organic layer was concentrated in vacuum, finally the solid mass was recrystallised from Ethanol to obtain the white colored solid (6) (6.9 g, 75 %); mp 151-153 °C. IR (KBr). 3555, 3378, 2950, 1683, 1547, 1311, 1172, 986, 781 cm⁻¹. ¹H NMR (400 MHz, DMSO-d₆) δ 8.31 (d, J = 8.4 Hz, 2H), 8.15 (d, J = 8.8 Hz, 2H), 7.11-7.34 (m, 2H), 7.19

(d, $J = 6.8$ Hz, 2H), 6.57 (d, $J = 8.8$ Hz, 1H), 4.86 (d, $J = 6.4$ Hz, 1H), 3.39-3.51 (m, 3H), 3.04-3.18 (m, 2H), 1.95-2.01 (m, 1H), 1.25 (s, 9H), 0.85 (d, $J = 7.2$ Hz, 6H) ppm. ^{13}C NMR: 156.2, 150.4, 146.5, 138.0, 129.1, 128.2, 126.4, 126.1, 124.7, 80.2, 73.5, 58.5, 57.3, 50.1, 36.4, 26.1 19.5, 24.2. R_f : 0.45; (hexane: ethyl acetate, 7:3 v/v). $[\alpha]_D^{25} = -27.7^\circ$ ($c = 1$ in CHCl_3). LCMS (m/z): 526[M+H] $^+$.

Synthesis of 4-Amino-N-((2R,3S)-3-amino-3-(4-fluorophenyl)-2-hydroxypropyl)-N-isobutylbenzenesulfonamide (10):

Tert-butyl (1S,2R)-1-(4-fluorophenyl)-2-hydroxy-3-(*N*-isobutyl-4-nitrophenylsulfonamido)propylcarbamate (9) (5.9 g, 0.01 mol), Fe-HCl (0.59 g) were suspended in methanol and the reaction mixture was taken into round bottom flask (RBF), rendered inert and evacuated. At an inside temperature of 45 °C, during 12 h, then catalyst was removed by filtration, progress of this reaction was monitored by TLC (methylenedichloride: methanol, 9:1). Concentrated HCl (3.7 mL) was added to the filtrate at room temperature, heated to reflux and stirred for 2 h. After complete conversion, pH of the reaction mixture was adjusted to 9.5 with 20% NaOH solution, most of the methanol was removed by distillation to get the crude product, it was recrystallised from a mixture of ethanol and water (40%). It was yielded 4.2 g, 96% of an off-white powder (7); mp 171-173 °C. R_f : 0.62; (methylenedichloride: methanol, 9:1 v/v): IR (KBr). 3452, 3352, 3240, 2957, 1651, 1600, 1321, 1149, 969, 959, 761 cm^{-1} . ^1H NMR (400 MHz, DMSO- d_6) δ 7.40 (d, $J = 8.4$ Hz, 2H), 7.24-7.34 (m, 2H), 7.16 (d, $J = 6.8$ Hz, 2H), 6.61 (d, $J = 8.4$ Hz, 2H), 5.95 (s, 2H), 4.62 (d, $J = 5.2$ Hz, 1H), 3.49 (brs, 1H), 3.31-3.38 (m, 1H), 2.69-2.91 (m, 3H), 2.91 (s, 1H), 2.33-2.36 (t, 1H), 1.91-1.95 (t, 1H), 1.22 (brs, 2H), 0.79- 0.83 (m, 6H) ppm. ^{13}C NMR: 150.2, 138.5, 130.0, 129.2, 128.2, 127.5, 125.6, 115.2, 78.2, 60.5, 58.2, 50.4, 38.9, 25.5, 18.3. $[\alpha]_D^{25} = -23.6^\circ$ ($c = 1$ in DMF). LCMS (m/z): 396[M+H] $^+$.

(3R, 3aS, 6aR)-Hexahydrofuro [2,3-*b*]furan-3-yl 4-nitrophenyl carbonate (7a) was prepared from (3R, 3aS, 6aR)-hexahydrofuro [2,3-*b*]furan-3-ol according to reported method.^{38, 51}

Synthesis of (3S,3aR,6aS)-Hexahydrofuro[2,3-*b*]furan-3-yl (1S,2R)-3-(4-amino-N-isobutylphenylsulfonamido)-1-(4-fluorophenyl)-2-hydroxypropylcarbamate (1): (3S, 3aR, 6aS)-Hexahydrofuro[2,3-*b*]furan-3-yl 4-nitrophenyl carbonate (1.96 g, 0.01 mol) (11)⁵¹ in methylene dichloride (15 mL) was added to a solution of 4-Amino-N-((2R,3S)-3-amino-3-(4-fluorophenyl)-2-hydroxypropyl)-N-isobutylbenzenesulfonamide (10) (2.6 g, 0.01 mol) and methylene dichloride (10 mL) at 0 °C during 2 h. The resulting mixture was raised to room temperature and stirred for 8 h. After completion of the reaction, monitored by HPLC, the reaction mixture was

quenched into mixture of water (26 mL) and ethyl acetate (30 mL), the aqueous layer was further extracted with ethyl acetate (2x15 mL), both organic layers were mixed and washed with 10% Na_2CO_3 solution (2x15 mL) followed by 10% NaCl solution (15 mL). The organic layer was dried over Na_2SO_4 , and concentrated in vacuum to get the off-white colored residue with 96% HPLC purity. It was purified by column chromatography by eluting with 5-10% methanol in methylene dichloride and concentrated in a rotary evaporator to get off-white crystals of 8 (3.26 g, 90%) with 99% HPLC purity; mp 116-118°C. IR (KBr): 3461, 3368, 3062, 2961, 1709, 1631, 1455, 1316, 1258, 1149, 1091, 833, 703 cm^{-1} . ^1H NMR (300 MHz, DMSO- d_6) δ 7.39 (d, $J = 17.4$ Hz, 2H), 7.24-7.34 (m, 2H), 7.16 (d, $J = 6.8$ Hz, 2H), 7.13-7.29 (m, 1H), 6.60 (d, $J = 17.4$ Hz, 2H), 5.96 (br s, 2H), 5.54 (d, $J = 10.2$ Hz 1H), 4.94 (d, $J = 12.6$ Hz, 1H), 4.77-4.84 (m, 1H 3.49 (br s, 1H), 3.54-3.84 (m, 1H), 3.54-3.84 (m, 1H), 3.54-3.84 (m, 1H), 3.34-3.41 (m, 1H), 3.34-3.41 (m, 1H), 3.25-3.26 (m, 1H), 2.87-3.03 (m, 1H), 2.87-3.03 (m, 1H), 2.57-2.78 (m, 1H), 2.57-2.78 (m, 1H), 1.86-1.96 (m, 1H), 1.86-1.96 (m, 1H), 1.69-1.72 (m, 1H), 0.84 (d, $J = 13.2$ Hz 6H) ppm. ^{13}C NMR: 155.18, 152.67, 139.43, 129.12, 128.96, 127.86, 125.71, 123.70, 112.62, 108.62, 72.16, 72.16, 70.29, 68.71, 57.03, 55.77, 52.52, 44.64, 35.14, 26.36, 25.51, 20.01. $[\alpha]_D^{25} = -22.8^\circ$ ($c = 1$ in CHCl_3). LC-MSD: 574 [M+Na] $^+$, 552 [M+H] $^+$. The enantiomeric ratio (% ee) was determined to be 99% by HPLC using Intersil ODS 3V 250*4.65 μm column (30% ethanol/ hexane + 1mL diethyl amine, 1 mL/min, 265 nm), t_R (minor, 8.33 min), t_R (major, 13.72 min).

CONCLUSION

In conclusion, we have developed first time a straightforward and novel synthetic methodology for (3S,3aR,6aS)-Hexahydrofuro [2,3-*b*]furan-3-yl (1S,2R)-3-(4-amino-N-isobutylphenylsulfonamido)-1-(4-fluorophenyl)-2-hydroxypropylcarbamate 1 from a readily available starting material, optically active L-phenylglycine 2, through the key intermediate *tert*-butyl formate (S, S)-oxiran-2-yl(phenyl)methanamine 7 which was obtained as an exclusive product without its stereoisomer. Darunavir analogue 1 was prepared on a gram scale in an overall yield of 43% with only one column chromatography separation in the eight-step synthesis.

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